STATISTICAL MOLECULAR THERMODYNAMICS

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Video 4.7

Ideal Polyatomic Gases: Part 1

ENERGY OF AN IDEAL POLYATOMIC GAS

In addition to translational and electronic degrees of freedom, a polyatomic also can rotate and has *multiple* vibrations.

The Energy:

$$\varepsilon_{\text{polyatomic}} = \varepsilon_{\text{trans}} + \varepsilon_{\text{rot}} + \varepsilon_{\text{vib}} + \varepsilon_{\text{elec}}$$
 (sum)

As for monatomic and diatomic gases, *translational* energy comes from the particle in a box approximation and depends only on the mass of the particle and a chosen volume

As for a diatomic gas, we assume a ground *electronic* state but instead of a single D_e we sum over the dissociation energies of *all* of the bonds

POLYATOMIC ROTATIONS

Linear polyatomics: Result the same as that for the diatomic rigid rotator. $q_{\rm rot} = \frac{T}{\sigma \Theta_{\rm rot}}$ $\sigma = 1 \text{ for COS}$ $\sigma = 2 \text{ for CO}_2$

Nonlinear polyatomics: For each of the 3 degrees of rotational freedom, (labeled A, B, C) we have a *separate moment of inertia* and a *separate rotational temperature*,

Three possible cases:

$$I_{A} = I_{B} = I_{C}, \quad \Theta_{rot,A} = \Theta_{rot,B} = \Theta_{rot,C} \quad \text{spherical top}$$

$$I_{A} = I_{B} \neq I_{C}, \quad \Theta_{rot,A} = \Theta_{rot,B} \neq \Theta_{rot,C} \quad \text{symmetric top}$$

$$I_{A} \neq I_{B} \neq I_{C}, \quad \Theta_{rot,A} \neq \Theta_{rot,B} \neq \Theta_{rot,C} \quad \text{asymmetric top}$$

NONLINEAR ROTATIONAL PARTITION FUNCTIONS

Quantum mechanics provides energy levels, for example,

for the spherical top:
$$\varepsilon_J = \frac{J(J+1)\hbar^2}{2I}$$
 $g_J = (2J+1)^2$ $J = 0, 1, 2, ...$

In all cases,

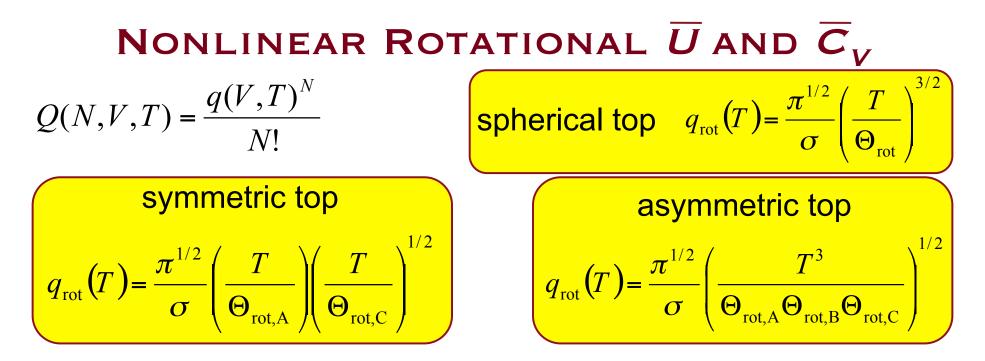
$$q_{\rm rot}(T) = \sum_{J=0}^{\infty} g_J e^{-\beta \varepsilon}$$

As long as each $\Theta_{rot} \ll T$, one can approximate ∞ the sum as an integral, $q_{rot} = \int_{\alpha}^{\infty} dJ g_J e^{-\beta \varepsilon_J}$

Which provides solutions:

spherical top
$$q_{rot}(T) = \frac{\pi^{1/2}}{\sigma} \left(\frac{T}{\Theta_{rot}}\right)^{3/2}$$

symmetric top
 $q_{rot}(T) = \frac{\pi^{1/2}}{\sigma} \left(\frac{T}{\Theta_{rot,A}}\right) \left(\frac{T}{\Theta_{rot,C}}\right)^{1/2}$
 $q_{rot}(T) = \frac{\pi^{1/2}}{\sigma} \left(\frac{T^{3}}{\Theta_{rot,A}\Theta_{rot,B}\Theta_{rot,C}}\right)^{1/2}$



Note that all $q_{\rm rot}$ have 3/2 power temperature dependence

• Energy
$$\overline{U}_{rot} = RT^2 \frac{\partial \ln T^{3/2}}{\partial T} = \frac{3}{2}RT$$

• Heat Capacity
$$\overline{C}_V = \left(\frac{\partial \overline{U}}{\partial T}\right)_V = \frac{3}{2}R$$