

STATISTICAL MOLECULAR THERMODYNAMICS

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Video 8.6

Natural Independent Variables

WORKING WITH THE SIMPLEST FORMS


$$dU = TdS - PdV \quad (\text{first and second laws})$$

If we consider S and V as independent variables of U , the coefficients of dS and dV are *simple* thermodynamic functions.

$$dU = \left(\frac{\partial U}{\partial S}\right)_V dS + \left(\frac{\partial U}{\partial V}\right)_S dV \quad \longrightarrow \quad \left(\frac{\partial U}{\partial S}\right)_V = T \quad \left(\frac{\partial U}{\partial V}\right)_S = -P$$

Compare with, say, V and T as independent variables:

$$dU = \left(\frac{\partial U}{\partial V}\right)_T dV + \left(\frac{\partial U}{\partial T}\right)_V dT$$

see Video 8.3 

$$= \left[T \left(\frac{\partial P}{\partial T}\right)_V - P \right] dV + C_V dT \quad \text{considerably more complex}$$

We thus refer to S and V as the *natural independent variables* of U

DIFFERENTIALS TO DATE

Start from the first and second laws: $dU = TdS - PdV$

Add $d(PV)$ to both sides: $\longrightarrow dH = TdS + VdP$

$$d(U+PV) = TdS - PdV + VdP + PdV$$

Subtract $d(TS)$ from both sides: $\longrightarrow dA = -SdT - PdV$

$$d(U-TS) = TdS - PdV - TdS - SdT$$

Add $d(PV)$ and subtract $d(TS)$ from both sides: $\longrightarrow dG = -SdT + VdP$

$$d(U+PV-TS) = TdS - PdV + VdP + PdV - TdS - SdT$$

(All from the 1st and 2nd laws and definitions of H , A , and G)

NATURAL INDEPENDENT VARIABLES

<u>Function</u>	<u>Differential</u>	<u>Variables</u>
U	$dU = TdS - PdV$	\longrightarrow S and V
S	$dS = \frac{1}{T}dU + \frac{P}{T}dV$	\longrightarrow U and V
H	$dH = TdS + VdP$	\longrightarrow S and P
A	$dA = -SdT - PdV$	\longrightarrow T and V
G	$dG = -SdT + VdP$	\longrightarrow T and P

ASSOCIATED MAXWELL RELATIONS

<u>Function</u>	<u>Differential</u>	<u>Maxwell relation</u>
U	$dU = TdS - PdV$	$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial P}{\partial S}\right)_V$
H	$dH = TdS + VdP$	$\left(\frac{\partial T}{\partial P}\right)_S = \left(\frac{\partial V}{\partial S}\right)_P$
A	$dA = -SdT - PdV$	$\left(\frac{\partial S}{\partial V}\right)_T = \left(\frac{\partial P}{\partial T}\right)_V$
G	$dG = -\overset{\downarrow}{S}d\overset{\downarrow}{T} + \overset{\downarrow}{V}d\overset{\downarrow}{P}$ <div style="display: flex; justify-content: center; gap: 20px; margin-top: -10px;"> ↑ ↑ ↑ ↑ </div>	$\overset{\downarrow}{\left(\frac{\partial S}{\partial P}\right)}_T = -\overset{\downarrow}{\left(\frac{\partial V}{\partial T}\right)}_P$ <div style="display: flex; justify-content: center; gap: 20px; margin-top: -10px;"> ↑ ↑ ↑ ↑ </div>