Name:	:	

### CHEMISTRY 5021/8021 MIDTERM EXAM — SPRING 2004

### 1. Molecular Mechanics (50 points)

Write two different functional forms that might be used to describe the potential energy associated with bond stretching. Carefully define all terms in the equations. Now, contrast the two: what are the relative merits of one compared to the other? Why might a given force field adopt one and not the other? Note that there are many possible answers here—I am not looking for any *specific* two forms.

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## 2. Semiempirical Theory (50 points)

An AM1 calculation on benzene provides a 298 K heat of formation of 23.5 kcal mol<sup>-1</sup>. Explain in a detailed fashion exactly how this heat of formation value was arrived at. You need *not* explain anything about geometry optimization.

# 3. Ab Initio MO Theory

a) How many basis functions are required for a HF calculation on glycine ( $C_2H_5NO_2$ ) using the 6-31G(d) basis set? (25 points)

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following questions. There is only one correct answer to each question. (5 points each)					
(1)	<u> </u>	evel employs second-order per th electron correlation and for			
(2)		neory expresses the complete ation state functions but fails t			
(3)		m with unrestricted Hartree-Foopen-shell Hartree-Fock wave	ock wave functions that is not functions.		
(4)	) This is a multilevel model that attempts to correct for incompleteness in basis set size and accounting for electron correlation by additive means.				
(5)	5) This single level of theory scales less favorably than MP3 but more favorably than MP5.				
	MP3	B: spin contamination	C: nondynamical correlation		
	G2 CISD	E: CCSD(T) H: ROHF	F: MP2 I: hyperconjugation		