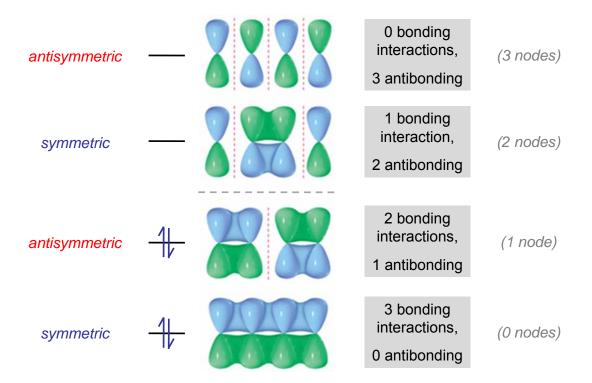
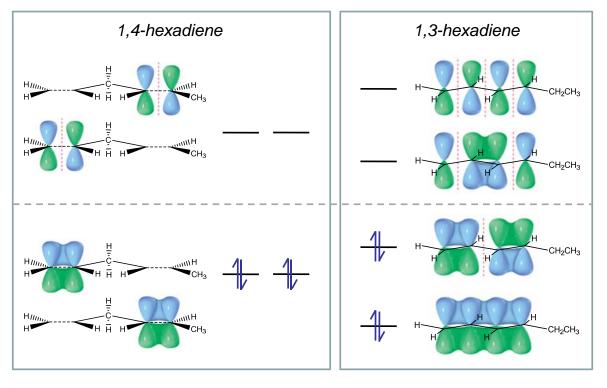
Conjugated Molecular Orbitals

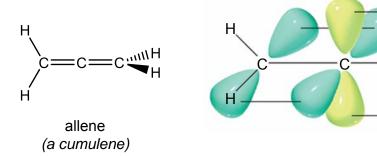


Conjugated Dienes Are More Stable Than Unconjugated Dienes



Diene Conformations

Cumulated Double Bonds Are Not Stabilized

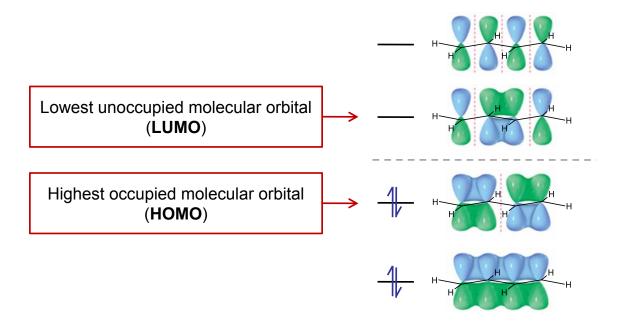


Central C is sp-hybridized, so it contributes 2 p orbitals to separate, perpendicular π -bonds.

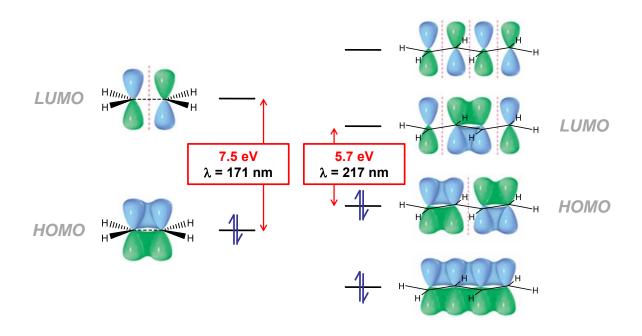
C:"IIIH

No orbital overlap, so orbital mixing between π -systems is not possible.

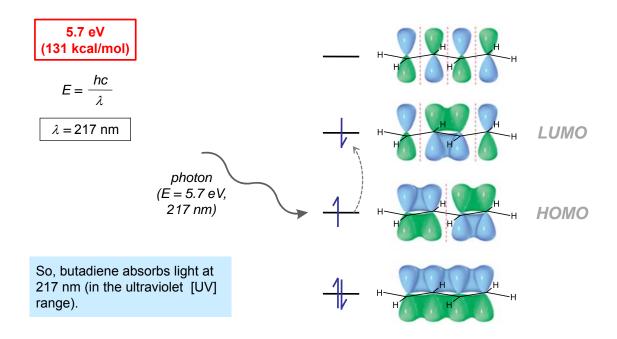
Conjugation Reduces the "HOMO-LUMO Gap"



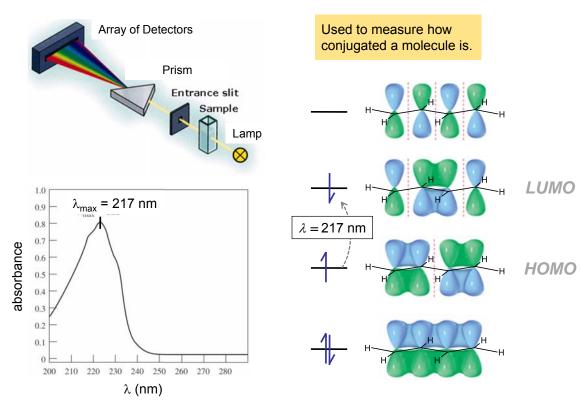
Conjugation Shifts UV-Vis Absorption to Longer Wavelengths



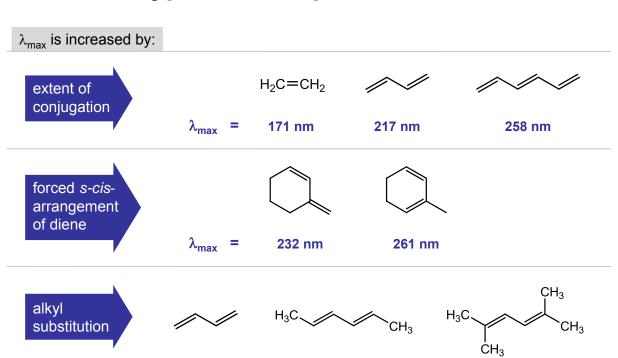
Electrons Can Be Promoted to Excited States By Light



UV/Vis Spectroscopy



Typical Absorption Maxima

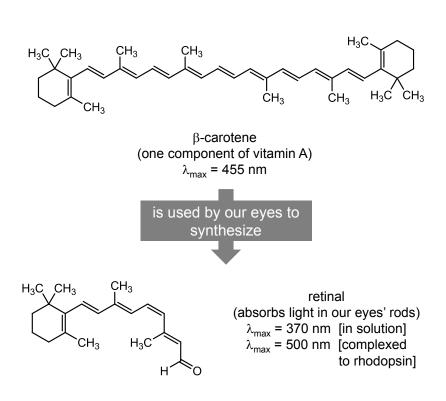


Typical Absorption Maxima

227 nm

217 nm

241 nm

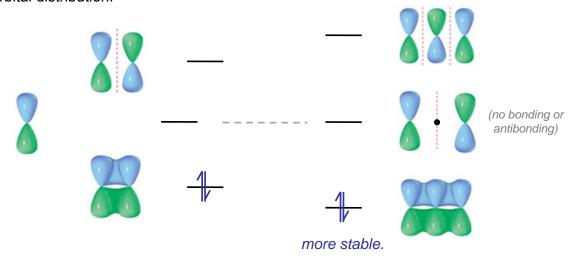


Conjugation Stabilizes Allylic Systems

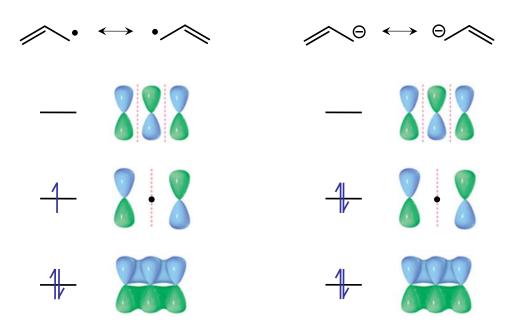
Resonance explanation: Stabilization of allyl cation is due to distribution of charge.

$$\bigoplus \bigoplus \bigoplus \bigoplus = \delta^{+} \bigwedge^{\delta^{+}}$$

Molecular orbital explanation: Stabilization also caused by orbital distribution.



Conjugation Stabilizes Allylic Systems



Reactions that go through allylic intermediates are usually accelerated (because the intermediates are more stable).