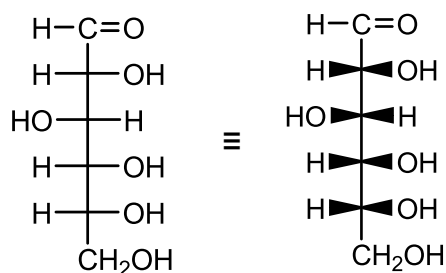


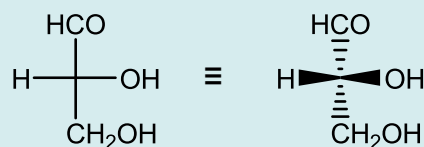
Carbohydrates

Molecules with formula $C_nH_{2n}O_n$.



D-*glucose*
($C_6H_{12}O_6$)

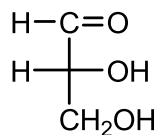
Fisher projections:



D-/L- notation: After placing most-oxidized carbon on top, sugars with last chiral $-\text{OH}$ on right are D-sugars.

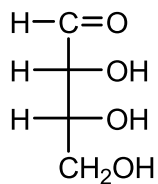
Monosaccharides: Aldoses and Ketoses

Aldoses are polyhydroxy-aldehydes.



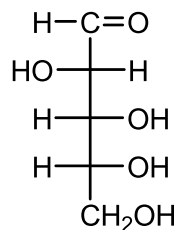
D-*glyceraldehyde*
($C_3H_6O_3$)

an aldotriose



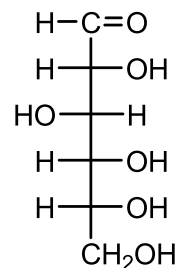
D-*erythrose*
($C_4H_8O_4$)

an aldotetrose



D-*arabinose*
($C_5H_{10}O_5$)

an aldopentose

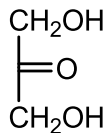


D-*glucose*
($C_6H_{12}O_6$)

an aldohexose

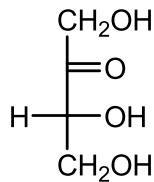
Monosaccharides: Aldoses and Ketoses

Ketoses are polyhydroxy-ketones.



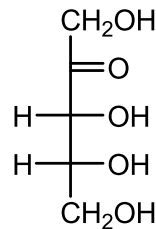
dihydroxyacetone
(C₃H₆O₃)

a ketotriose



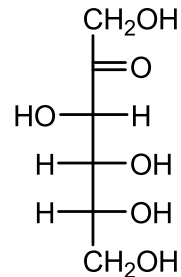
D-erythrulose
(C₄H₈O₄)

a ketotetrose



D-ribulose
(C₅H₁₀O₅)

a ketopentose

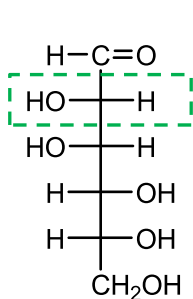


D-fructose
(C₆H₁₂O₆)

a ketohexose

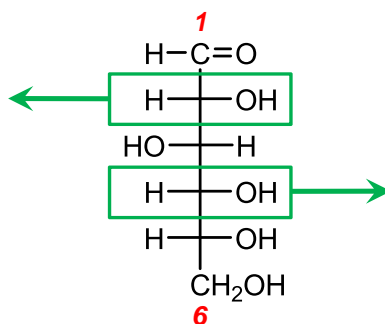
Epimers

Epimers differ in stereochemistry at one carbon.

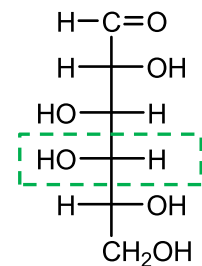


D-mannose

the **C2**-epimer of
glucose



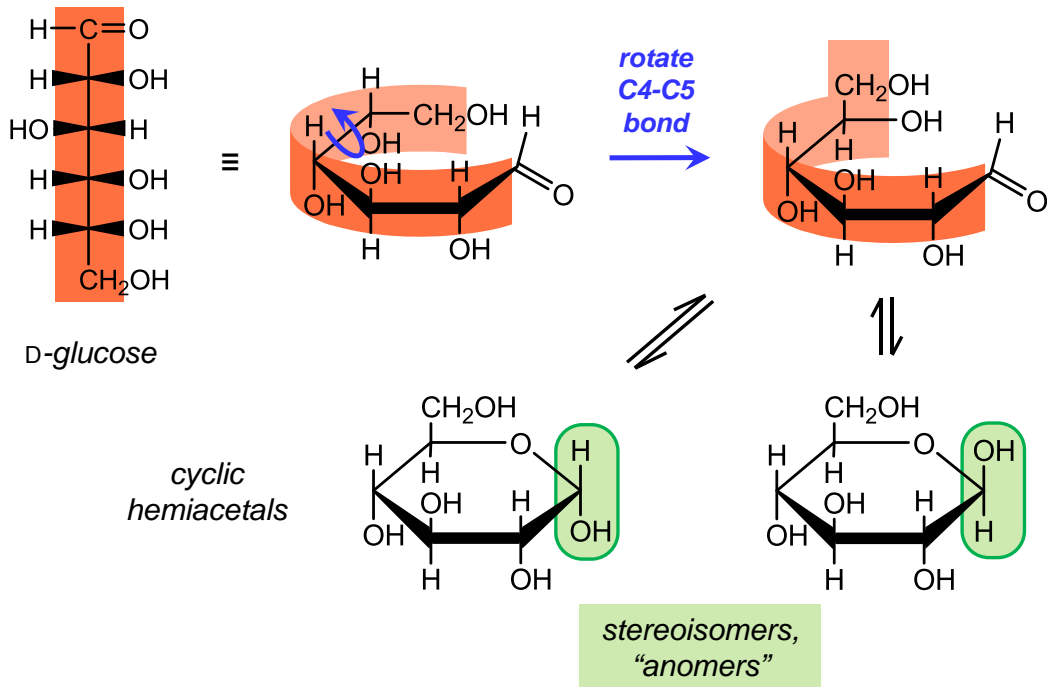
D-glucose



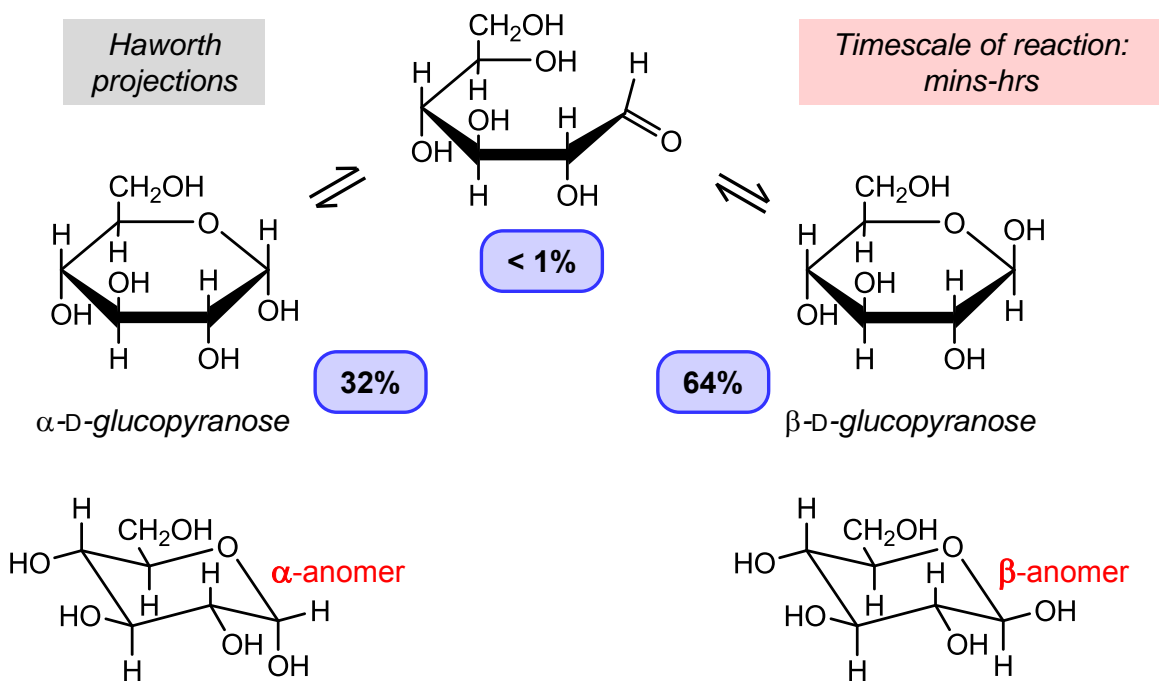
D-galactose

the **C4**-epimer of
glucose

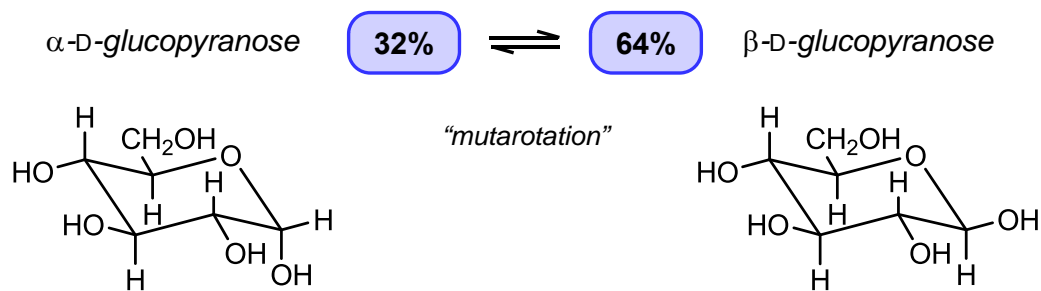
Monosaccharides Equilibrate Between Acyclic and Cyclic Forms



Equilibrating Glucose Anomers

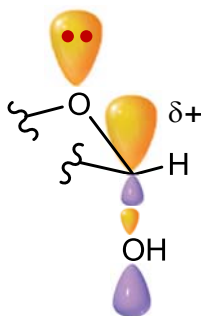


The Anomeric Effect



Advantage:
The anomeric effect.

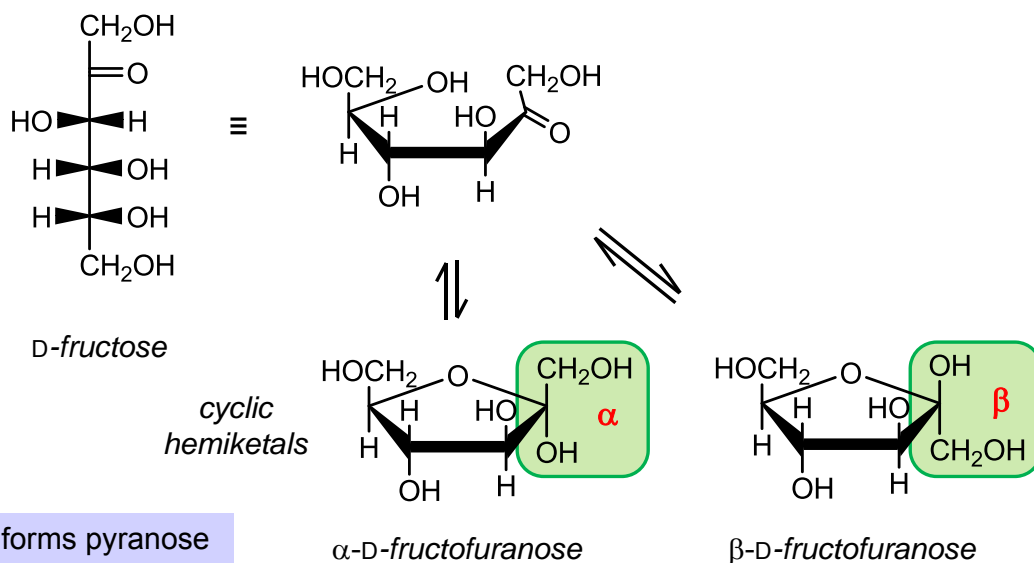
Interaction between O_{sp^3} lone pair and C1-O σ^* antibonding orbital is stabilizing, and orbitals are oriented correctly.



Advantage:
Hydroxyl group is equatorial.

Anomeric effect not possible for β -anomer; orbitals don't line up.

Ketoses Also Equilibrate Between Acyclic and Cyclic Forms



Also forms pyranose (6-membered) rings.

70%

30%

, but found most often in complex sugars.