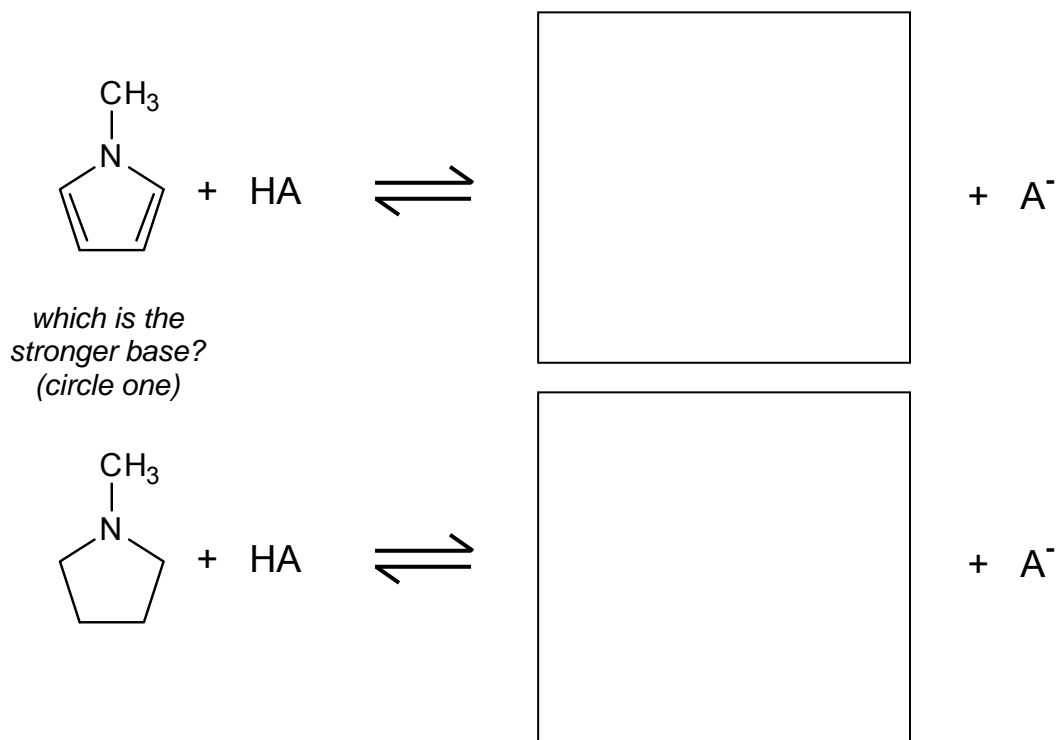


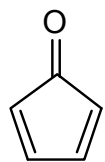
Workshop 5
Aromatic Ions in Acid-Base Reactions

For each of the pairs of bases below,

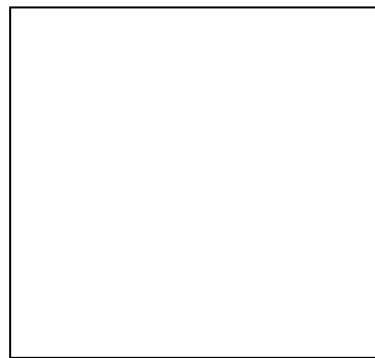
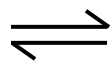
- Draw the conjugate acid in the box; and
- Predict which will be the stronger base.



(continued on next page)

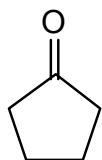


+ HA

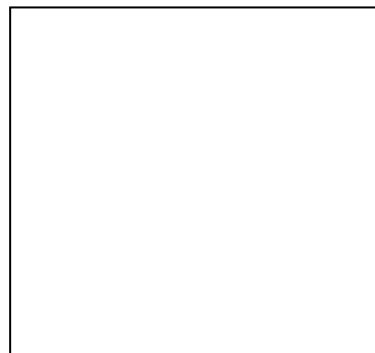
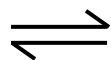


+ A⁻

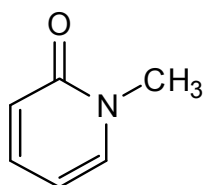
*which is the stronger base?
(circle one)*



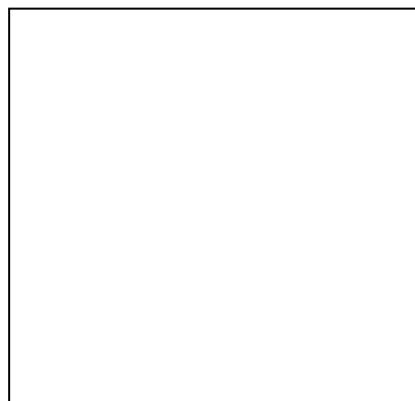
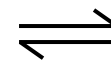
+ HA



+ A⁻

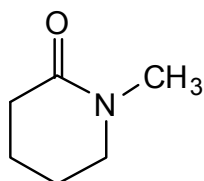


+ HA

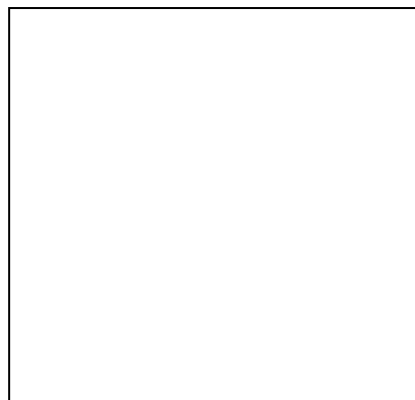


+ A⁻

*which is the stronger base?
(circle one)*



+ HA



+ A⁻