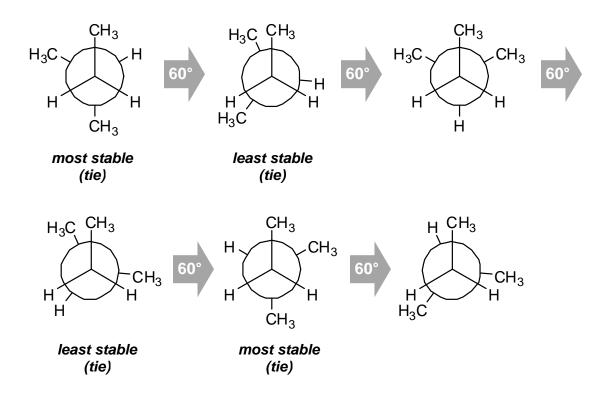
## Chemistry 2331H

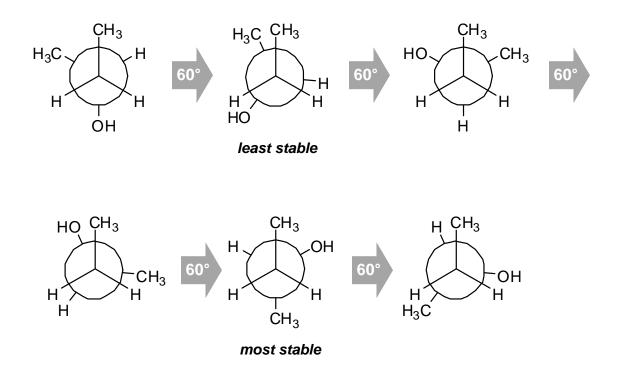
## In-Class Exercise Solutions Newman Projections

1. Rotating the group in the back,



Staggered conformations with only one gauche interaction are the most stable. Eclipsed conformations with one totally eclipsed methyl-methyl interaction are the least stable.

2. Again rotating the group in the back,



-OH is slightly less repelling than  $-CH_3$ , and  $-OH/-CH_3$  eclipsing interactions are not as bad as  $-CH_3/-CH_3$  interactions, because (i) the  $\sigma$ -bond to H can swing away from the opposing  $-CH_3$ , and (ii) the O lone pairs that would then swing towards the  $-CH_3$ are held more closely to the O nucleus than a bonding pair would be, and thus don't repel across the divide as much. So there are no ties in this case.

