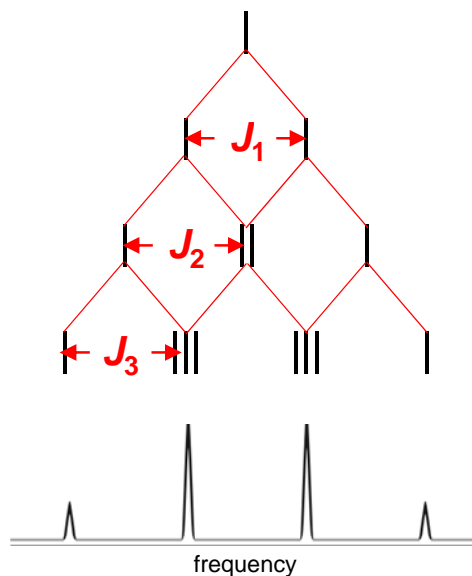
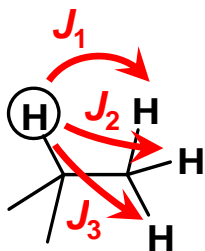


Spin-Spin Coupling: Simple Splitting and Pascal's Triangle

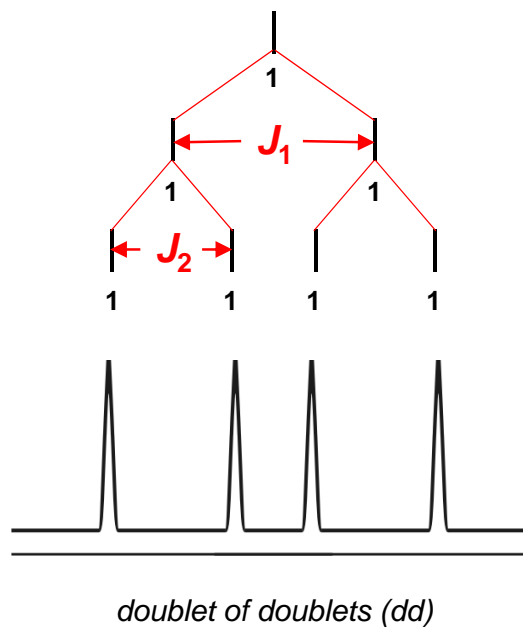
Can think of Pascal's triangle as a way of expressing splitting for multiple coupling constants J_1 , J_2 , J_3 , etc., where $J_1 = J_2 = J_3$.



Spin-Spin Coupling: More Complex Splitting

What if coupling constants aren't equal?

Splitting pattern gets more complicated, doesn't follow Pascal's triangle.

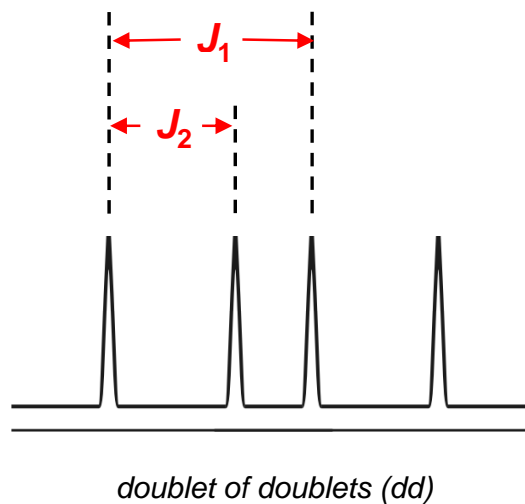


Spin-Spin Coupling: More Complex Splitting

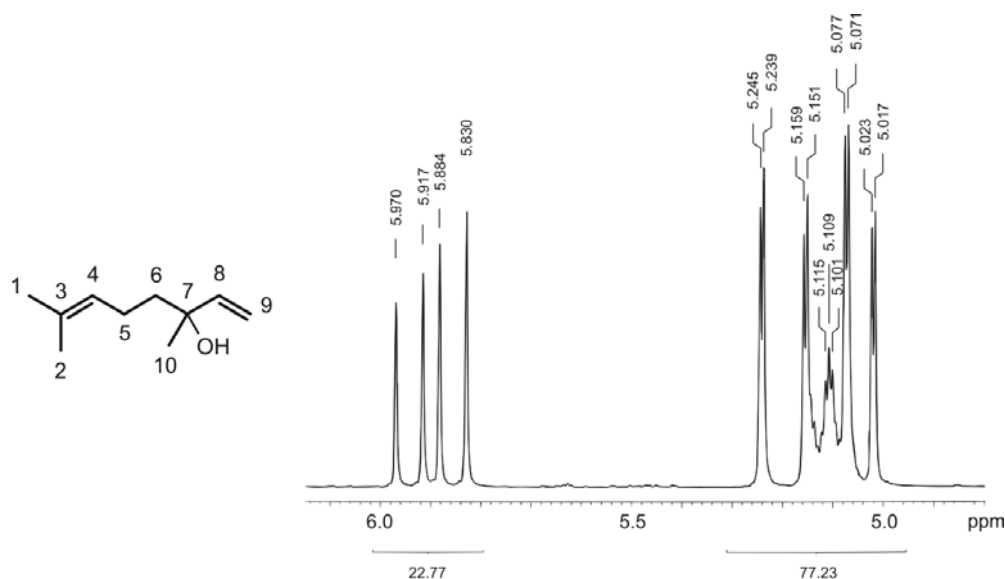
What if coupling constants aren't equal?

Splitting pattern gets more complicated, doesn't follow Pascal's triangle.

For case of doublet of doublets, can measure J 's directly from spectrum.



Spin-Spin Coupling: More Complex Splitting



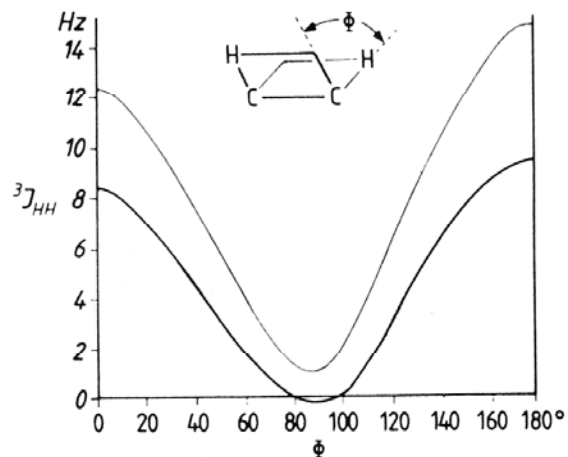
200 MHz ¹H NMR spectrum of linalool, 5-6 ppm region.

Nuclear Coupling and Geometry: The Karplus Equation

For vicinal coupling partners,

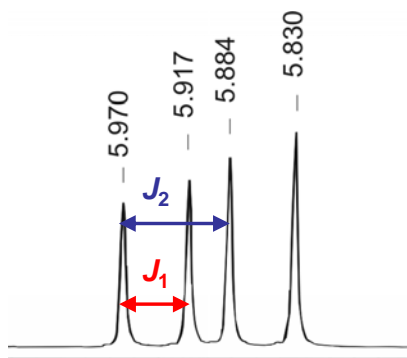
$${}^3J_{\text{H-H}} = a(\cos^2\phi) + c$$

$$a(0-90^\circ) < a(90-180^\circ)$$



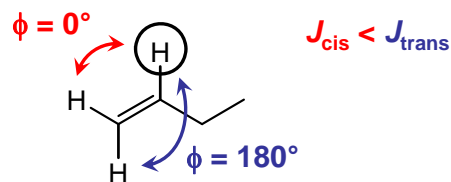
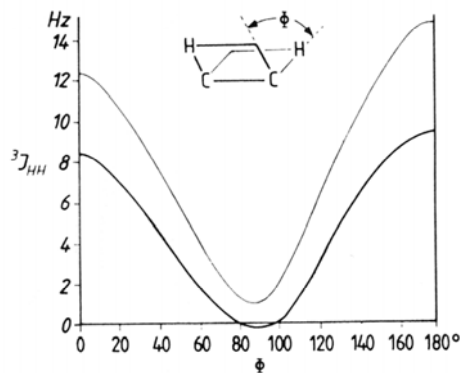
Nuclear Coupling and Geometry: The Karplus Equation

From spectrum of linalool...



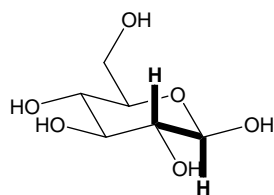
$$J_1 = 10.6 \text{ Hz}$$

$$J_2 = 17.2 \text{ Hz}$$

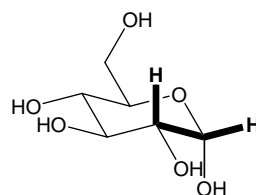


Applying the Karplus Equation

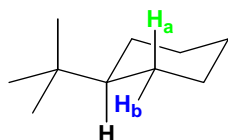
Other examples:



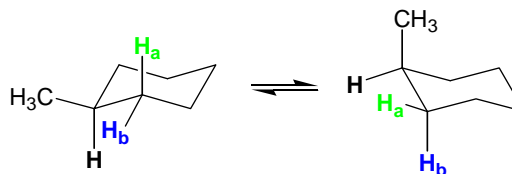
$$J_{\alpha\text{-glucose}} = 7.8 \text{ Hz}$$



$$J_{\beta\text{-glucose}} = 3.6 \text{ Hz}$$



$$J_{\text{H-Ha}} = 12 \text{ Hz} \quad J_{\text{H-Hb}} = 4 \text{ Hz}$$



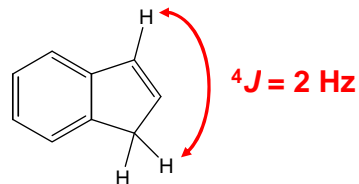
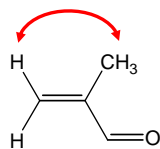
$$J_{\text{H-Ha}} = 9 \text{ Hz} \quad J_{\text{H-Hb}} = 4 \text{ Hz}$$

$$J_{\text{TOT}} = f_{\text{ax-ax}}(J_{\text{ax-ax}}) + f_{\text{eq-eq}}(J_{\text{eq-eq}})$$

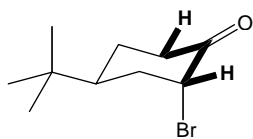
Long-Range Coupling

Occurs through multiple bonds or "W" conformation.

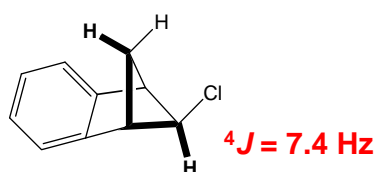
$${}^4J = 1.5 \text{ Hz}$$



$${}^4J = 2 \text{ Hz}$$



$${}^4J = 3 \text{ Hz}$$



$${}^4J = 7.4 \text{ Hz}$$